Domino Reactions

DOI: 10.1002/ange.201205899

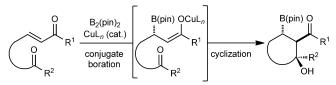
Enantioselective Copper(I)-Catalyzed Borylative Aldol Cyclizations of Enone Diones**

Alan R. Burns, Jorge Solana González, and Hon Wai Lam*

The in situ trapping of metal enolates generated by the catalytic hydrometalation of α,β -unsaturated carbonyl compounds has proven to be a mild and versatile method for carbon–carbon bond construction. Not only does this strategy allow precise control of the site of enolization of substrates containing several acidic sites, the use of chiral metal/ligand complexes can also enable products to be formed with high levels of diastereo- and enantiocontrol. In this context, we and others have developed various copper(I)-catalyzed or copper(I)-mediated reductive aldol, Annich, and Michael Reactions to furnish products with high diastereo- and enantioselectivities.

While these processes are effective, the development of related transformations in which metal enolate generation is initiated not by the formation of a carbon–hydrogen bond, but by a carbon–heteroatom bond which can then be exploited in subsequent functionalizations, should also be of high value. Given the recent developments in enantioselective copper(I)-catalyzed conjugate boration reactions, we envisaged a domino conjugate boration/aldol cyclization sequence in which a copper enolate generated from the initial conjugate boration is trapped by a pendant ketone to deliver cyclic products containing multiple stereocenters (Scheme 1).

Relevant precedent for such a process is relatively limited. [10] Hoveyda and co-workers have described racemic



Scheme 1. Proposed conjugate boration/aldol cyclization.

[*] Dr. A. R. Burns, J. Solana González, Dr. H. W. Lam EaStCHEM, School of Chemistry, University of Edinburgh Joseph Black Building, The King's Buildings, West Mains Road, Edinburgh EH9 3JJ (UK)

E-mail: h.lam@ed.ac.uk

Homepage: http://homepages.ed.ac.uk/hlam/index.html

[**] We thank the EPSRC for financial support and for a Leadership Fellowship (to H.W.L.). We thank Darryl W. Low (University of Edinburgh) for assistance in the preparation of substrates, and Gary S. Nichol (University of Edinburgh) for assistance with X-ray crystallography. Solvias AG and AllyChem are gratefully acknowledged for kind donations of chiral ligands and B₂(pin)₂, respectively. We thank the EPSRC National Mass Spectrometry Service Centre for providing high-resolution mass spectra.



Supporting information for this article is available on the WWW under http://dx.doi.org/10.1002/anie.201205899.

N-heterocyclic-carbene-catalyzed conjugate boration of a cyclic enone with subsequent in situ trapping of the resulting boron enolate with benzaldehyde, [10a] whereas Kanai, Shibasaki, and co-workers reported a similar enantioselective process catalyzed by a chiral copper/bisphosphine complex. [10b] Recently, Riant and co-workers developed a racemic copper-catalyzed conjugate boration/intermolecular aldol sequence of acyclic or cyclic α , β -unsaturated carbonyl compounds. [10c] However, analogous processes in which the aldol reaction occurs in an intramolecular fashion have not, to our knowledge, been reported, despite the potential for the generation of useful functionalized cyclic building blocks.

In view of the precedent set by Krische and co-workers, who described highly diastereo- and enantioselective rho-dium-catalyzed conjugate arylation/aldol cyclizations of enone diones,^[11] we wondered whether a related process involving copper-catalyzed conjugate boration could be developed. Herein, we report the enantioselective copper-catalyzed domino conjugate boration/aldol cyclization of enone diones to give a range of highly functionalized bicyclic products.^[12-17] This desymmetrization process results in the formation of one boron–carbon bond, one carbon–carbon bond, and four contiguous stereocenters, two of which are quaternary, with high levels of diastereo- and enantioselection.

Our studies commenced with evaluation of various common chiral bisphosphine ligands (L1-L4; 5.5 mol%) in the domino conjugate boration/aldol cyclization of the enone dione 1a with B₂(pin)₂ (1.1 equiv) in the presence of CuCl (5 mol %) and NaOtBu (7.5 mol %) in THF (0.1M) at room temperature (Table 1, entries 1-4). Although the desired bicyclic product 3a was formed in greater than 95:5 d.r. [(major isomer):(Σ other isomers)] in several cases, appreciable quantities of the product 2a resulting from conjugate boration without cyclization were often observed (Table 1, entries 1, 2, and 4). With the Taniaphos ligand L3, 2a was the sole product (Table 1, entry 3). Regarding enantioinduction, the Josiphos ligand L4^[8c,d] provided the best results, furnishing 3a as the major product (3a/2a = 93:7) in 85% ee (Table 1, entry 4).[18] Additional optimization revealed that the inclusion of MeOH^[8c,d] (2.0 equiv; Table 1, entry 5) and halving the concentration of the reaction (Table 1, entry 6) led to further increases in enantioselectivity, albeit with a lower ratio of 3a/ 2a (Table 1, entry 6). Fortunately, the use of more-hindered alcohol additives improved this ratio without sacrificing enantioselectivity (Table 1, entries 7 and 8), with iPrOH providing the best results (entry 7). It should be noted that there is the potential for enantioenrichment of 3a through a ligand-controlled double asymmetric process, in which the copper enolate containing the minor stereogenicity formed in



Table 1: Optimization of reaction conditions. [a]

Entry	Ligand	Conc [M]	ROH	Conv [%] ^[b]	3 a / 2 a ^[b]	3 a ee [%] ^[c]
1	L1	0.1	_	> 95	86:14	70
2	L2	0.1	_	> 95	31:69	16 ^[d]
3	L3	0.1	_	56	< 5:95	_[e]
4	L4	0.1	-	> 95	93:7	85
5	L4	0.1	MeOH	> 95	87:13	90
6	L4	0.05	MeOH	> 95	72:28	95
7	L4	0.05	<i>i</i> PrOH	> 95	87:13	94 ^[f]
8	L4	0.05	tBuOH	>95	84:16	94

[a] Reactions were conducted using 1a (0.20 mmol). [b] Determined by H NMR spectroscopy of the unpurified reaction mixture. [c] Determined by HPLC analysis using a chiral stationary phase. [d] Opposite enantiomer of 3 a obtained. [e] Enantiomeric excess of 2 a was 6%. [f] Uncyclized product 2a was isolated in 12% yield with 93% ee. pin = pinacolato, Cy = cyclohexyl.

the initial conjugate boration either does not cyclize, or cyclizes to form an epimeric product. However, 2a and 3a were formed with similar ee values (93% ee for 2a and 94% ee for 3a) from the experiment of Table 1, entry 7, which suggests that a double asymmetric process does not occur to any significant extent in this case.

As shown in Scheme 2, a range of enone diones underwent borylative aldol cyclization under the optimized reaction conditions to furnish [6,6]-bicyclic products with high levels of diastereoselection (>95:5 d.r.) and enantioselection $(\geq 92\% ee)$. [18] In addition to the original substrate **1a**, which gave 3a in 71% yield and 95% ee, substrates containing a 4chlorophenyl ketone or a 4-methoxyphenyl ketone performed well (products 3b and 3c). Substrate 1d, containing an allyl group rather than a methyl group at the 2-position of the cyclic 1,3-diketone, was also a competent substrate (product **3d**). The process is not limited to α,β -unsaturated aromatic ketones; the reactions of substrates 1e and 1f, which contain an aliphatic ketone and an α,β -unsaturated ester, respectively, were also highly stereoselective (products 3e and 3f). The reaction may also be performed on a larger scale. The reaction of 1b on a 3.15 mmol scale provided 1.06 g (77% yield) of 3b in 93% ee.

The reaction of substrate 1g, which contains a 4-nitrophenyl ketone, was poorly diastereoselective and gave two separable products (Scheme 3). The major product 3ga was

Scheme 2. Borylative aldol cyclizations of enone diones to form decalins. The reactions were performed with 1 a-f (0.30 mmol) in THF (6 mL). Yields are of pure isolated major diastereomers. Diastereomeric ratios [(major isomer):(Σ other isomers)] were determined by ¹H NMR analysis of the unpurified reaction mixtures. Enantiomeric excesses were determined by HPLC analysis using a chiral stationary phase, and refer to the major diastereomer. [a] Values in parentheses refer to a reaction conducted using 3.15 mmol of 1b. [b] Reaction conducted with tBuOH (2.0 equiv) in place of iPrOH.

Scheme 3. Borylative aldol cyclizations of substrates 1 g-i.

isolated in 49% yield and 86% ee, but interestingly, the relative stereochemistry of this compound differed from those products depicted in Scheme 2.[18] The originally expected product 3gb was the minor component in this reaction (29% yield, 87% ee). Two diastereomeric products were also isolated from the reactions of substrates 1h and 1i, which contain a 3-(trifluoromethyl)phenyl and a 2-bromophenyl ketone, respectively.

Plausible reactive conformations that rationalize the diastereochemical outcomes of these reactions are illustrated in Figure 1. The formation of the products **3a–f**, **3gb**, **3hb**, and

Figure 1. Models to explain diastereoselectivity of the reactions.

3ib may be explained by aldol cyclization through a Z enolate in the chairlike Zimmerman–Traxler-type transition-state^[19] **TS1**, in which the B(pin) substituent occupies a pseudoequatorial position in the tether connecting the dione and enolate components. The formation of **3ga**, **3ha**, and **3ia** may be accounted for on the basis of cyclization through an E enolate in the chairlike transition-state **TS2**, or through a E enolate in the twist-boat transition-state **TS3**. However, the reasons that determine the propensity of a particular substrate to react through one or more of the possible pathways are not clear at the present time.

The scope of the enantioselective borylative cyclization was additionally extended to the preparation of products containing different ring sizes (Scheme 4). In these cases, the use of *t*BuOH in place of *i*PrOH proved to be marginally superior. Pleasingly, [5,6]-, [6,5]-, and [5,5]-bicyclic ring systems were formed with good yields and high levels of enantioinduction (92–99% *ee* for the major diastereomer).^[18] As with the decalin systems (Scheme 2), the major diastereomers for the [5,6]-hydrindanes, **5a–d**, have the *syn-syn-anti* configuration shown, as confirmed by X-ray crystallographic analysis of **5d**.^[18] Lower levels of diastereoselection were observed with products **5e** and **5f**, which were isolated as inseparable mixtures of diastereomers.

Next, the parallel kinetic resolution^[21] of a racemic chiral enone-1,3-dicarbonyl substrate was examined.^[11] Under standard conditions using *i*PrOH as the alcohol additive, the substrate **6**, containing a cyclic β -ketoamide, provided the tricycle **7** in 42 % yield, 83:17 d.r., and 91 % *ee* (for the major diastereomer), along with the uncyclized product **8** in 41 % yield, 73:27 d.r., and 90 % *ee* (for the major diastereomer) [Eq. (1)]. ^[22] The formation of each of the products **7** and **8** as an inseparable diastereomeric mixture indicates that the parallel kinetic resolution was only moderately effective for this substrate. When the reaction was performed with the achiral ligand P(OEt)₃, **7** and **8** were obtained in diastereo-

Scheme 4. Borylative aldol cyclizations of enone diones to form hydrindanes and diquinanes. The reactions were performed with 4a–f (0.30 mmol) in THF (6 mL). Yields are of isolated material. Diastereomeric ratios [(major isomer):(Σ other isomers)] were determined by ^1H NMR analysis of the unpurified reaction mixtures. Enantiomeric excesses were determined by HPLC analysis using a chiral stationary phase. [a] Isolated as an inseparable mixture of diastereomers. [b] Enantiomeric excess values in parentheses refer to the minor diastereomer, the stereochemistry of which was not assigned unambiguously.

meric ratios similar to those indicated in Equation (1), thus suggesting that the diastereoselectivity of the cyclization



event is governed predominantly by the substrate. However, the high levels of enantioselectivity obtained for products **7** and **8** in Equation (1) suggests this process is still potentially useful.

To demonstrate the utility of the cyclization products, further transformations of the decalin **3b** were conducted (Scheme 5). Treatment of **3b** with NaBO₃·H₂O^[23] provided

Scheme 5. Further transformations of product 3 b.

the secondary alcohol **9** in 86% yield. Furthermore, formation of the potassium trifluoroborate salt **10** was readily achieved by reaction of **3b** with KHF₂.^[24]

In summary, we have described highly enantioselective copper(I)-catalyzed borylative aldol cyclizations of enone diones that result in densely functionalized decalin-, hydrindane-, and diquinane-based products containing four contiguous stereocenters, two of which are quaternary. Work to develop additional enantioselective metal-catalyzed domino processes is underway in our laboratories.

Received: July 24, 2012 Published online: September 28, 2012

Keywords: asymmetric catalysis · boron · copper · domino reactions · enantioselectivity

- [1] R. R. Huddleston, M. J. Krische, Synlett 2003, 12-21.
- For relevant reviews of reductive aldol reactions, see Ref. [1] and: a) H. Nishiyama, J.-i. Ito, Chem. Commun. 2010, 46, 203–212; b) S. B. Han, A. Hassan, M. J. Krische, Synthesis 2008, 2669–2679; c) S. A. Garner, S. B. Han, M. J. Krische, Modern Reduction Methods, Wiley-VCH, Weinheim, 2008, pp. 387–417; d) H. Nishiyama, T. Shiomi, Top. Curr. Chem. 2007, 279, 105–137; e) M.-Y. Ngai, J.-R. Kong, M. J. Krische, J. Org. Chem. 2007, 72, 1063–1072; f) H.-Y. Jang, M. J. Krische, Eur. J. Org. Chem. 2004, 3953–3958; g) H.-Y. Jang, M. J. Krische, Acc. Chem. Res. 2004, 37, 653–661; h) W. B. Motherwell, Pure Appl. Chem. 2002, 74, 135–142.
- [3] For intramolecular copper-catalyzed or copper-mediated reductive aldol reactions, see: a) P. Chiu, C.-P. Szeto, Z. Geng, K.-F. Cheng, Org. Lett. 2001, 3, 1901–1903; b) P. Chiu, S. K. Leung, Chem. Commun. 2004, 2308–2309; c) H. W. Lam, P. M. Joensuu, Org. Lett. 2005, 7, 4225–4228; d) H. W. Lam, G. J. Murray, J. D.

- Firth, *Org. Lett.* **2005**, *7*, 5743–5746; e) B. H. Lipshutz, B. Amorelli, J. B. Unger, *J. Am. Chem. Soc.* **2008**, *130*, 14378–14379; f) J. Deschamp, O. Riant, *Org. Lett.* **2009**, *11*, 1217–1220; g) N. Li, J. Ou, M. Miesch, P. Chiu, *Org. Biomol. Chem.* **2011**, *9*, 6143–6147; h) J. Ou, W.-T. Wong, P. Chiu, *Tetrahedron* **2012**, *68*, 3450–3456; i) J. Deschamp, T. Hermant, O. Riant, *Tetrahedron* **2012**, *68*, 3457–3467.
- [4] For intermolecular copper-catalyzed reductive aldol reactions, see: a) T. Ooi, K. Doda, D. Sakai, K. Maruoka, *Tetrahedron Lett.* 1999, 40, 2133–2136; b) D. Zhao, K. Oisaki, M. Kanai, M. Shibasaki, *Tetrahedron Lett.* 2006, 47, 1403–1407; c) A. Welle, S. Diez-Gonzalez, B. Tinant, S. P. Nolan, O. Riant, *Org. Lett.* 2006, 8, 6059–6062; d) O. Chuzel, J. Deschamp, C. Chausteur, O. Riant, *Org. Lett.* 2006, 8, 5943–5946; e) J. Deschamp, O. Chuzel, J. Hannedouche, O. Riant, *Angew. Chem.* 2006, 118, 1314–1319; *Angew. Chem. Int. Ed.* 2006, 45, 1292–1297; f) D. Zhao, K. Oisaki, M. Kanai, M. Shibasaki, *J. Am. Chem. Soc.* 2006, 128, 14440–14441.
- [5] For enantioselective copper-catalyzed reductive coupling of alkenylazaarenes with ketones, see: A. Saxena, B. Choi, H. W. Lam, J. Am. Chem. Soc. 2012, 134, 8428–8431.
- [6] Y. Du, L.-W. Xu, Y. Shimizu, K. Oisaki, M. Kanai, M. Shibasaki, J. Am. Chem. Soc. 2008, 130, 16146–16147.
- [7] a) T. M. Kamenecka, L. E. Overman, S. K. Ly Sakata, *Org. Lett.* 2001, 4, 79–82; b) C. L. Oswald, J. A. Peterson, H. W. Lam, *Org. Lett.* 2009, 11, 4504–4507.
- [8] For seminal references in this area, see: a) K. Takahashi, T. Ishiyama, N. Miyaura, *Chem. Lett.* 2000, 29, 982 983; b) H. Ito, H. Yamanaka, J.-i. Tateiwa, A. Hosomi, *Tetrahedron Lett.* 2000, 41, 6821 6825; c) S. Mun, J.-E. Lee, J. Yun, *Org. Lett.* 2006, 8, 4887 4889; d) J.-E. Lee, J. Yun, *Angew. Chem.* 2008, 120, 151 153; *Angew. Chem. Int. Ed.* 2008, 47, 145 147.
- [9] For selected recent reviews, see: a) E. Hartmann, D. J. Vyas, M. Oestreich, *Chem. Commun.* 2011, 47, 7917–7932; b) J. A. Schiffner, K. Müther, M. Oestreich, *Angew. Chem.* 2010, 122, 1214–1216; *Angew. Chem. Int. Ed.* 2010, 49, 1194–1196; c) V. Lillo, A. Bonet, E. Fernández, *Dalton Trans.* 2009, 2899–2908.
- [10] a) K.-S. Lee, A. R. Zhugralin, A. H. Hoveyda, J. Am. Chem. Soc.
 2009, 131, 7253-7255; b) I. H. Chen, L. Yin, W. Itano, M. Kanai,
 M. Shibasaki, J. Am. Chem. Soc. 2009, 131, 11664-11665; c) A.
 Welle, V. Cirriez, O. Riant, Tetrahedron 2012, 68, 3435-3443.
- [11] B. M. Bocknack, L.-C. Wang, M. J. Krische, Proc. Natl. Acad. Sci. USA 2004, 101, 5421–5424.
- [12] For copper-catalyzed reductive aldol cyclizations to form similar products, see Refs. [3f-i].
- [13] For rhodium-catalyzed reductive aldol cyclizations to form similar products, see: a) R. R. Huddleston, M. J. Krische, *Org. Lett.* 2003, 5, 1143–1146; b) P. K. Koech, M. J. Krische, *Org. Lett.* 2004, 6, 691–694.
- [14] For copper-catalyzed conjugate addition/cyclization using organozinc reagents to form similar products, see: K. Agapiou, D. F. Cauble, M. J. Krische, J. Am. Chem. Soc. 2004, 126, 4528–4529.
- [15] For Lewis acid promoted cyclizations to form similar products, see: a) T. L. Macdonald, C. M. Delahunty, K. Mead, D. E. O'Dell, *Tetrahedron Lett.* 1989, 30, 1473-1476; b) A. Balog, S. V. Geib, D. P. Curran, *J. Org. Chem.* 1995, 60, 345-352; c) D. Schinzer, G. Panke, *J. Org. Chem.* 1996, 61, 4496-4497; d) C. E. Davis, R. M. Coates, *Angew. Chem.* 2002, 114, 509-511; *Angew. Chem. Int. Ed.* 2002, 41, 491-493; e) B. Ressault, A. Jaunet, P. Geoffroy, S. Goudedranche, M. Miesch, *Org. Lett.* 2012, 14, 366-369.
- [16] For radical-promoted cyclizations to form similar products, see: a) G. A. Molander, Y. L. Huerou, G. A. Brown, J. Org. Chem. 2001, 66, 4511–4516; b) A. Beauseigneur, C. Ericsson, P. Renaud, K. Schenk, Org. Lett. 2009, 11, 3778–3781.
- [17] For fluoride-promoted cyclizations to form similar products, see: a) C. Kuroda, K. Ito, Bull. Chem. Soc. Jpn. 1996, 69, 2297 – 2303;

- b) D. Schinzer, T. Blume, P. G. Jones, *Angew. Chem.* 1996, 108, 2669–2671; *Angew. Chem. Int. Ed. Engl.* 1996, 35, 2500–2502;
 c) P. Geoffroy, M. P. Ballet, S. Finck, E. Marchioni, M. Miesch, *Tetrahedron* 2010, 66, 7012–7016.
- [18] The relative and absolute stereochemistries of the products herein were assigned by analogy to those of **3b**, **3ia**, and **5d**, which were determined by X-ray crystallography. See the Supporting Information for full details. CCDC 891571 (**3b**), 891572 (**3ia**), and 891573 (**5d**) contain the supplementary crystallographic data for this paper. These data can be obtained free of charge from The Cambridge Crystallographic Data Centre via www.ccdc.cam.ac.uk/data_request/cif.
- [19] H. E. Zimmerman, M. D. Traxler, J. Am. Chem. Soc. 1957, 79, 1920–1923.

- [20] D. A. Evans, C. W. Downey, J. T. Shaw, J. S. Tedrow, Org. Lett. 2002, 4, 1127 – 1130.
- [21] For a review of parallel kinetic resolution, see: J. R. Dehli, V. Gotor, *Chem. Soc. Rev.* **2002**, *31*, 365–370.
- [22] The *ee* values of the major and minor isomers of **8** were obtained on the secondary alcohols resulting from oxidation of the boronate ester. See the Supporting Information for details.
- [23] C. N. Farthing, S. P. Marsden, *Tetrahedron Lett.* 2000, 41, 4235–4238.
- [24] For reviews of organotrifluoroborates, see: a) G. A. Molander, D. L. Sandrock, *Curr. Opin. Drug Discovery Dev.* 2009, 12, 811 823; b) S. Darses, J.-P. Genet, *Chem. Rev.* 2008, 108, 288 325; c) H. A. Stefani, R. Cella, A. S. Vieira, *Tetrahedron* 2007, 63, 3623 3658.